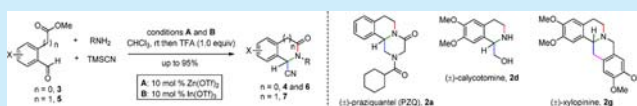


Approach to Isoindolinones, Isoquinolinones, and THIQs via Lewis Acid-Catalyzed Domino Strecker-Lactamization/Alkylations

Sivasankaran Dhanasekaran,^{†,§} Arun Suneja,^{‡,§} Vishnumaya Bisai,^{‡,¶} and Vinod K. Singh^{*,†,‡}[‡]Department of Chemistry, Indian Institute of Science Education and Research, Bhopal, MP - 462 066, India[†]Department of Chemistry, Indian Institute of Technology, Kanpur, UP - 208 016, India

S Supporting Information

ABSTRACT: A one-pot, three-component synthesis of widely substituted isoindolinones and isoquinolinones, featuring a Lewis acid-catalyzed efficient Strecker reaction and lactamization sequence, affording products in good to high yields is reported. The method has also been extended to the synthesis of tetrahydroisoquinolines (THIQs) in high yields.



Isoindolinones are an important class of heterocycles found in many biologically active natural products¹ and are also useful intermediates in the synthesis of a variety of drug molecules² (1a–b; Figure 1) having important biological

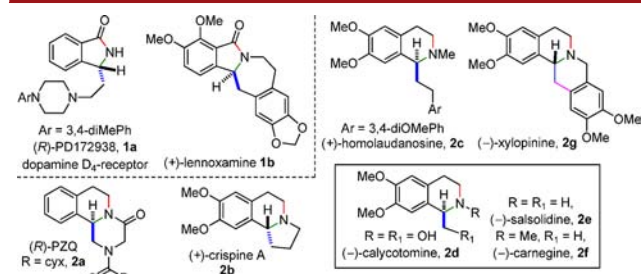


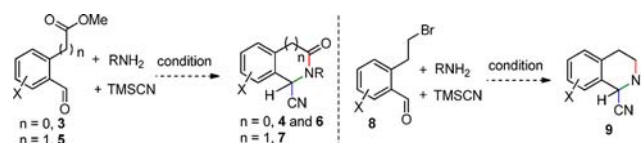
Figure 1. Selected active isoindolinones and THIQs.

activities such as antihypertensive,³ antipsychotic,⁴ anti-inflammatory,⁵ anesthetic,⁶ antiulcer,⁷ vasodilatory,⁸ antiviral,⁹ and antileukemic¹⁰ agents. In addition, a few members of this class are known to have platelet aggregation inhibitory activity¹¹ and are shown to induce dose-dependent p53-dependent gene transcription in MDM2-amplified SJSA human sarcoma cell lines.¹² On the other hand, tetrahydroisoquinolines¹³ (THIQs) (2a–g; Figure 1) are widespread in nature and have interesting biological activities. They are also found to be the building blocks for synthesis of many complex alkaloids.¹⁴

Approaches in literature for isoindolinone synthesis include Heck cyclization,¹⁵ Diels–Alder approach,¹⁶ ring-closure of hydrazones,¹⁷ reactions of acyliminium ion,¹⁸ exploiting carbanion methodology,¹⁹ and various enantioselective approaches²⁰ including our recent domino²¹ Cu(I)-catalyzed enantioselective propargylation/lactamization and organocatalytic Mannich/lactamization sequence.²² On the other hand, synthesis of isoquinolines involves various diastereoselective processes^{13c,23} and catalytic enantioselective processes.^{21,24} Although some elegant approaches to these targets have been reported, still there exist a fewer number of reports to

synthesize them applying a general catalytic route. To this end, recently, we reported the Lewis acid-catalyzed allylation-lactamization domino process²⁵ for synthesis of diversely substituted isoindolinones and THIQs. Herein, we report a practical approach to isoindolinones, isoquinolinones, and THIQs via a Lewis acid-catalyzed domino Strecker-lactamization/alkylations of readily available *o*-formyl methylbenzoates, *o*-formyl methylarylacetaes, and *o*-formyl aryethyl bromides (Scheme 1).

Scheme 1. Proposed Strecker-Lactamization/Alkylations



At the outset, optimization of domino Strecker-lactamization was performed by taking 1.0 equiv of each ester-aldehyde 3a and PMPNH₂ (*p*-anisidine) with 1.5 equiv of TMSCN without any catalyst. We found that only Strecker product 10a was formed in 28–42% yields, and no trace of isoindolinone 4a were observed (Scheme 2). Following extensive optimization, it was found that 10 mol % of Zn(OTf)₂ and In(OTf)₃ each afforded 4a in 93% yield (condition A) and 91% yield (condition B), respectively, when 1.0 equiv of TFA was used (see the SI for details).

Scheme 2. Domino Strecker-Lactamization Process

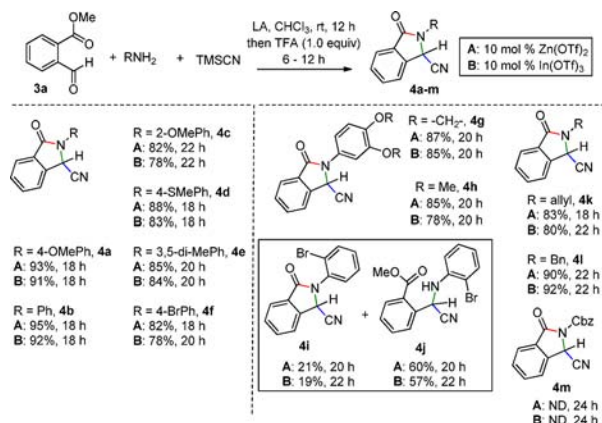


Received: November 19, 2015

Published: February 4, 2016

With optimized conditions, a variety of amines such as aromatic, aliphatic, and amine having an electron-withdrawing group (Cbz-NH₂) were studied. To our delight, various aromatic amines, such as aniline, *o*-methoxyphenyl (OMP), *p*-thiomethoxyphenyl (PTMP), 3,4-methylenedioxyphenyl, 3,4-dimethoxyphenyl, 3,5-dimethylphenyl, and *p*-bromophenyl, all afforded expected Strecker-lactamization products **4b–h** in up to 95% yield (Scheme 3) under conditions A and B. It is

Scheme 3. Scope of Reaction with Various Amines^{a–c}



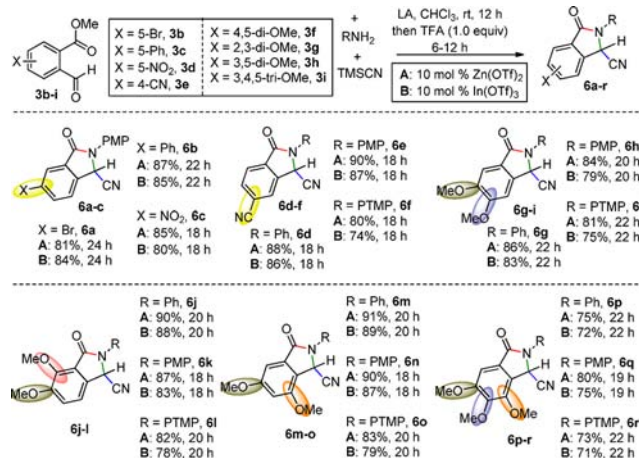
^aCondition A: 10 mol % Zn(OTf)₂. ^bCondition B: 10 mol % In(OTf)₃. ^c1.0 equiv of TFA was used after 12 h as an additive. LA = Lewis acid, TFA = trifluoroacetic acid.

important to note that isoindolinones with *N*-aryl groups having electron-donating groups (such as **4a**, **4c–d**, and **4g–h**) in the ring can be cleaved under oxidative conditions so as to obtain free amines.²⁶ To our surprise, in the case of *o*-bromophenylamine, we could only isolate the expected isoindolinone in 21% yield, where uncyclized benzylamine **4j** was obtained in 57–60% yields. We speculate that this could be because of the $-I$ effect of the $-Br$ group at the *ortho*-position of the aryl ring, which hampers the formation of isoindolinone (Scheme 3).²⁷

Then, we turned our attention to domino Strecker-lactamization using an aliphatic amine having easily removable groups such as allyl and benzyl. Gratifyingly, under optimized conditions we were able to synthesize isoindolinones **4k–l** having *N*-allyl and *N*-benzyl protecting groups in 80–92% yields. We also used CbzNH₂ as a protected amine source having an electron-withdrawing group;²⁸ however, to our displeasure, we were not able to assess expected isoindolinone **4m**, and this led to a multitude of spots on the TLC, indicating the nucleophilicity of amines plays a crucial role in the domino Strecker-lactamization process.

Further, we investigated the substrate scope of domino Strecker-lactamization using different amines *viz.* aniline (R = Ph), *p*-methoxyphenylamine (R = PMP), and *p*-thiomethoxyphenylamine (R = PTMP) with various *o*-formyl benzoates under optimized conditions A and B (Scheme 4). Gratifyingly, it was found that various *o*-formyl benzoates such as **3b–i** are suitable substrates under optimized conditions. In the presence of 10 mol % Zn(OTf)₂ (condition A) and In(OTf)₃ (condition B) followed by treatment with 1.0 equiv of TFA as an additive, these *o*-formyl benzoates afforded a variety of products having *N*-aryl groups in synthetically viable yields. Methyl *o*-formyl benzoates, having bromo, phenyl, and nitro substitution at the

Scheme 4. Substrate Scope of Isoindolinone Synthesis^{a–c}



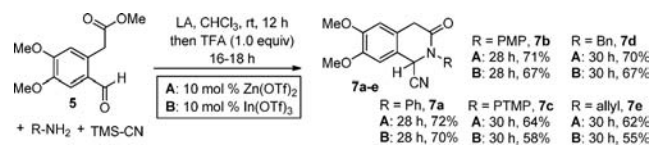
^aCondition A: 10 mol % Zn(OTf)₂. ^bCondition B: 10 mol % In(OTf)₃. ^c1.0 equiv of TFA was used after 12 h as additive. LA = Lewis acid, TFA = trifluoroacetic acid.

p-position of aldehyde such as **3b–d**, all afforded products **6a–c** in high yields (Scheme 4).

Methyl *o*-formyl benzoates having cyano substitution at the *p*-position of an ester such as **3e** also afforded products **6d–f** in good to excellent yields (Scheme 4). Importantly, less reactive methyl *o*-formyl benzoates having the $-OMe$ group as an electron-rich substituent at the *p*-positions of aldehyde, such as **3f–i**, also underwent a domino Strecker-lactamization sequence efficiently to afford **6g–r** in synthetically viable yields (Scheme 4).

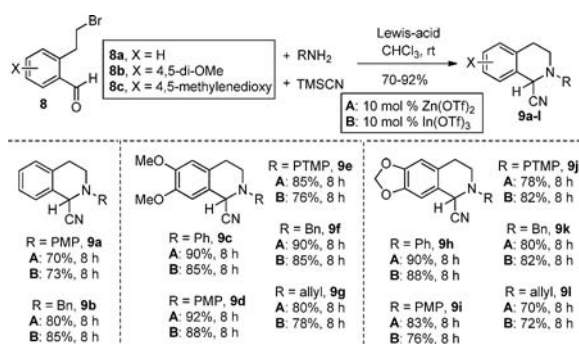
Having secured the synthesis of isoindolinones, we attempted a one-pot, three-component synthesis of isoquinolinones using various amine partners as per our hypothesis (Scheme 1). As most of the naturally occurring THIQ alkaloids contain the dimethoxy substituent, we explored the possibility of the formation of domino Strecker-lactamization using *o*-formyl methylaryllacetate **5** (Scheme 5). To our delight, we were able to synthesize a variety of isoquinolinones **7a–e** in synthetically useful yields, following our optimized conditions A and B (Scheme 5).

Scheme 5. Substrate Scope of Isoquinolinone Synthesis^{a–c}



^aCondition A: 10 mol % Zn(OTf)₂. ^bCondition B: 10 mol % In(OTf)₃. ^c1.0 equiv of TFA was used after 12 h as additive. LA = Lewis acid, TFA = trifluoroacetic acid.

Further, we thought to apply the domino Strecker-*N*-alkylation sequence for the synthesis of tetrahydroisoquinolines (THIQs). Toward this, a few *o*-formyl phenethyl bromides **8a–c** were employed in the domino Strecker-*N*-alkylation which afforded a variety of *N*-protected THIQs **9a–I** in 70–92% yields (Scheme 6). Interestingly, in these cases an additive (TFA) is not required, thus indicating S_N² type *N*-alkylation is a faster process than a lactamization (Schemes 3–5).

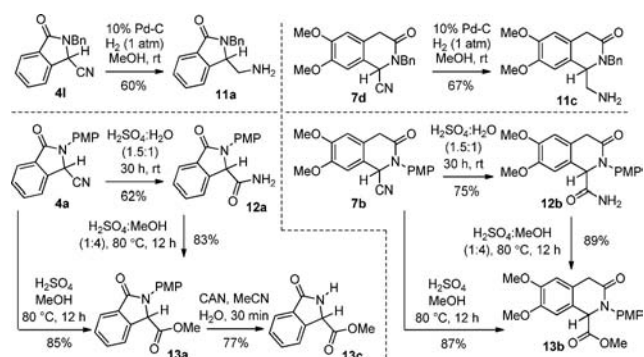
Scheme 6. Substrate Scope Using Various Bromo Aldehydes^{a,b}

^aCondition A: 10 mol % $\text{Zn}(\text{OTf})_2$. ^bCondition B: 10 mol % $\text{In}(\text{OTf})_3$.

The domino Strecker-lactamization was also carried out on a 1.0 g scale of *o*-formyl methylbenzoates **3a** and *o*-formyl arylacetate **5** under condition A, which afforded products **4a**, **7b**, and **7d** in 90%, 78%, and 93% yields, respectively (see the SI for details). Similarly, a gram scale reaction of bromoaldehyde **8b** with *p*-methoxyphenylamine and **8a** with benzylamine under condition A afforded THIQs **9d** and **9b** in 76% and 78% yields, respectively (see the SI for details).

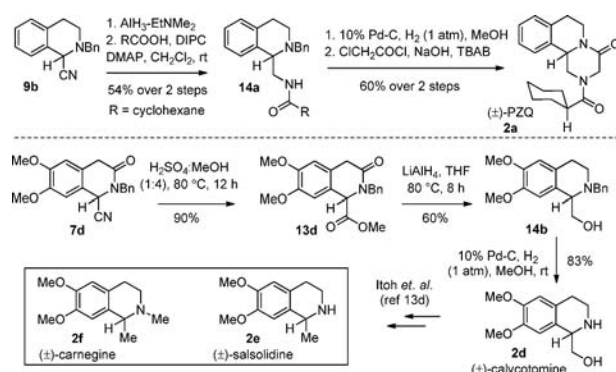
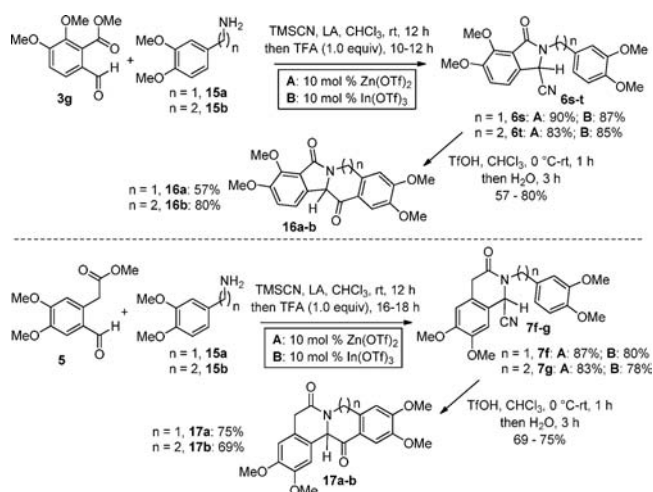
To show versatility, compounds **4l** and **7d** were treated under hydrogenolysis separately to obtain amine compounds **11a** and **11c** in 60% and 67% yields, respectively (Scheme 7),

Scheme 7. Synthetic Elaborations



where *N*-benzyl groups were found to be untouched. We also reacted compounds **4a** and **7b** in the presence of aqueous H_2SO_4 to afford carboxamides **12a** and **12b** in 62% and 75% yields, respectively (Scheme 7). Esters **13a** and **13b** can be obtained from **4a** and **7b**, respectively, when reactions were carried out in MeOH. Oxidative cleavage of the PMP group can simply be accomplished with CAN to afford isoindolinone **13c** (Scheme 7). Later, synthesis of the anthelmintic drug (\pm)-praziquantel **2a**²⁹ and alkaloid (\pm)-calycotomine **2d**^{13d} has been accomplished from **9b** and **7d** via intermediacy of **14a** and **13d**, respectively (Scheme 8).

Nevertheless, Lewis acid-catalyzed domino Strecker-lactamization was applied for the synthesis of isoindolinones **6s–t**, which on subsequent treatment with triflic acid (TfOH) afforded tetracyclic structures **16a–b** (Scheme 9), resembling the core structure of lennoxamine **1b** (Figure 1). Along similar lines isoquinolinones **7f–g** can also be utilized for the synthesis of tetracyclic structures **17a–b** (Scheme 9), which are common structural features of the protoberberine alkaloid, xylopinine

Scheme 8. Synthesis of (\pm)-PZQ **2a** and (\pm)-Calycotomine **2d**Scheme 9. Synthesis of Tetracyclic Structures **16a–b** and **17a–b**

(**2g**) (Figure 1). In these transformations, we have shown that Friedel–Crafts type C–C bond-forming reactions can be performed on nitrile functionality.

In conclusion, we report a Lewis acid-catalyzed highly efficient domino Strecker-lactamization for the synthesis of isoindolinones and isoquinolinones. The reaction itself is operationally simple and proceeds under mild conditions to afford isoindolinones with high yields. A variety of valuable intermediates have been synthesized utilizing the aforementioned method. Further exploration of the enantioselective version of this process is currently under active investigation.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b03331.

Experimental procedures and analytical data (^1H , ^{13}C NMR spectra and HRMS) for all new compounds (PDF)

CIF file of **4a** (CIF)

CIF file of **9d** (CIF)

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: vinodks@iitk.ac.in.

Present Address

[¶]Department of Chemistry, Indian Institute of Science Education and Research Tirupati, Tirupati - 517 507, Andhra Pradesh, India.

Author Contributions

[§]S.D. and A.S. contributed equally.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

Financial support through the SERB, DST (EMR/2014/001165) is gratefully acknowledged. We thank Ms. Anusha Upadhyay, IISER Bhopal for X-ray structure analysis. S.D. and A.S. thank the CSIR, New Delhi for fellowships (SRF).

■ REFERENCES

- (1) (a) Pigeon, P.; Decroix, B. *Tetrahedron Lett.* **1997**, 38, 2985. (b) Abramovitch, R. A.; Shinkai, I.; Mavunkel, B. J.; More, K. M.; O'Connor, S.; Ooi, G. H.; Pennington, W. T.; Srinivasan, P. C.; Stowers, J. R. *Tetrahedron* **1996**, 52, 3339.
- (2) (a) Boger, D. L.; Lee, J. K.; Goldberg, J.; Jin, Q. *J. Org. Chem.* **2000**, 65, 1467. (b) Egbertson, M. S.; Hartman, G. D.; Gould, R. J.; Bednar, R. A.; Cook, J. J.; Gaul, S. L.; Holahan, M. A.; Libby, L. A.; Lynch, J. J.; Sitko, G. R.; Stranieri, M. T.; Vassallo, L. M. *Bioorg. Med. Chem. Lett.* **1996**, 6, 2519.
- (3) Ferland, J.-M.; Demerson, C. A.; Humber, L. G. *Can. J. Chem.* **1985**, 63, 361.
- (4) (a) Linden, M.; Hadler, D.; Hofmann, S. *Hum. Psychopharmacol.* **1997**, 12, 445. (b) Zhuang, Z.-P.; Kung, M.-P.; Mu, M.; Kung, H. F. *J. Med. Chem.* **1998**, 41, 157.
- (5) Li, S.; Wang, X.; Guo, H.; Chen, L. *Yiyano Gongue* **1985**, 16, 543; *Chem. Abstr.* **1986**, 105, 6378n.
- (6) Laboratori Baldacci, S. P. A. Japanese Patent 5, 946, 268, 1984; *Chem. Abstr.* **1984**, 101, 54922.
- (7) (a) Lippmann, W. U.S. Patent 4,267,189, 1981; *Chem. Abstr.* **1981**, 95, 61988m. (b) Belliotti, T. R.; Brink, W. A.; Kesten, S. R.; Rubin, J. R.; Wustrow, D. J.; Zoski, K. T.; Whetzel, S. Z.; Corbin, A. E.; Pugsley, T. A.; Heffner, T. G.; Wise, L. D. *Bioorg. Med. Chem. Lett.* **1998**, 8, 1499.
- (8) Achinami, K.; Ashizawa, N.; Kobayasui, F. Japanese Patent 03,133,955, 1991; *Chem. Abstr.* **1991**, 115, 255977j.
- (9) (a) Pendrak, I.; Barney, S.; Wittrock, R.; Lambert, D. M.; Kingsbury, W. D. *J. Org. Chem.* **1994**, 59, 2623. (b) De Clercq, E. *J. Med. Chem.* **1995**, 38, 2491.
- (10) (a) Taylor, E. C.; Zhou, P.; Jennings, L. D.; Mao, Z.; Hu, B.; Jun, J.-G. *Tetrahedron Lett.* **1997**, 38, 521. (b) Riedinger, C.; Endicott, J. A.; Kemp, S. J.; Smyth, L. A.; Watson, A.; Valeur, E.; Golding, B. T.; Griffin, R. J.; Hardcastle, I. R.; Noble, M. E.; McDonnell, J. M. *J. Am. Chem. Soc.* **2008**, 130, 16038.
- (11) (a) Taylor, E. C.; Zhou, P.; Jennings, L. D.; Mao, Z.; Hu, B.; Jun, J.-G. *Tetrahedron Lett.* **1997**, 38, 521. (b) Fushka, J.; Fuskova, A.; Proksa, B. *Zb. Pr. Chemickotechnol Fak, SVST* **1979–1981**, No. pub. 1986, 285–291; *Chem. Abstr.* **1987**, 106, 95582k.
- (12) Riedinger, C.; Endicott, J. A.; Kemp, S. J.; Smyth, L. A.; Watson, A.; Valeur, E.; Golding, B. T.; Griffin, R. J.; Hardcastle, I. R.; Noble, M. E.; McDonnell, J. M. *J. Am. Chem. Soc.* **2008**, 130, 16038.
- (13) (a) Corey, E. J.; Gin, D. Y.; Kania, R. S. *J. Am. Chem. Soc.* **1996**, 118, 9202. (b) Flanagan, M. E.; Williams, R. M. *J. Org. Chem.* **1995**, 60, 6791. (c) Review: Chrzanowska, M.; Rozwadowska, M. D. *Chem. Rev.* **2004**, 104, 3341. (d) Kanemitsu, T.; Yamashita, Y.; Nagata, K.; Itoh, T. *Synlett* **2006**, 2006, 1595. (e) Dhanasekaran, S.; Kayet, A.; Suneja, A.; Bisai, V.; Singh, V. K. *Org. Lett.* **2015**, 17, 2780.
- (14) (a) Scott, J. D.; Williams, R. M. *J. Am. Chem. Soc.* **2002**, 124, 2951. (b) Wu, Y.-C.; Liron, M.; Zhu, J. *J. Am. Chem. Soc.* **2008**, 130, 7148 and references cited.
- (15) Grigg, R.; Dorrity, M. J. R.; Malone, J. F.; Mongkolaos-Savaratana, T.; Norbert, W. D. J. A.; Sridharan, V. *Tetrahedron Lett.* **1990**, 31, 3075.
- (16) McAlonan, H.; Murphy, J. P.; Nieuwenhuyzen, M.; Reynolds, K.; Sarma, P. K. S.; Stevenson, P. J.; Thompson, N. J. *Chem. Soc., Perkin Trans 1* **2002**, 69 and references therein.
- (17) (a) Enders, D.; Braig, V.; Raabe, G. *Can. J. Chem.* **2001**, 79, 1528. (b) Adachi, S.; Onozuka, M.; Yoshida, Y.; Ide, M.; Saikawa, Y.; Nakata, M. *Org. Lett.* **2014**, 16, 358.
- (18) (a) Bahajaj, A. A.; Vernon, J. M.; Wilson, G. D. *Tetrahedron* **2004**, 60, 1247. (b) Chen, M.-D.; Zhou, X.; He, M.-Z.; Ruan, Y.-P.; Huang, P.-Q. *Tetrahedron* **2004**, 60, 1651. (c) Opatz, T.; Ferenc, D. *J. Org. Chem.* **2004**, 69, 8496. (d) Sun, L. X.; Zeng, T.; Jiang, D.; Dai, L.-Y.; Li, C.-J. *Can. J. Chem.* **2012**, 90, 92.
- (19) (a) Comins, D. L.; Schilling, S.; Zhang, Y. *Org. Lett.* **2005**, 7, 95. (b) Pérard-Viret, J.; Prangé, T.; Tomas, A.; Royer, J. *Tetrahedron* **2002**, 58, 5103.
- (20) (a) Wang, Z.-Q.; Feng, C.-G.; Xu, M.-H.; Lin, G. Q. *J. Am. Chem. Soc.* **2007**, 129, 5336. (b) Guo, S.; Xie, Y.; Hu, X.; Xia, C.; Huang, H. *Angew. Chem., Int. Ed.* **2010**, 49, 2728. (c) Yang, G.; Shen, C.; Zahng, W. *Angew. Chem., Int. Ed.* **2012**, 51, 9141. (d) Tiso, S.; Palombi, L.; Vignes, C.; Mola, A. D.; Massa, A. *RSC Adv.* **2013**, 3, 19380.
- (21) Bisai, V.; Suneja, A.; Singh, V. K. *Angew. Chem., Int. Ed.* **2014**, 53, 10737.
- (22) Bisai, V.; Unhale, R. A.; Suneja, A.; Dhanasekaran, S.; Singh, V. K. *Org. Lett.* **2015**, 17, 2102.
- (23) (a) Endo, A.; Yanagisawa, A.; Abe, M.; Tohma, S.; Kan, T.; Fukuyama, T. *J. Am. Chem. Soc.* **2002**, 124, 6552. (b) Chen, J.; Chen, X.; Bois-Choussy, M.; Zhu, J. *J. Am. Chem. Soc.* **2006**, 128, 87.
- (24) (a) Kitamura, H.; Hsiao, Y.; Ohta, M.; Tsukamoto, M.; Ohta, T.; Takaya, H.; Noyori, R. *J. Org. Chem.* **1994**, 59, 297. (b) Uematsu, N.; Fujii, A.; Hashiguchi, S.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1996**, 118, 4916. (c) Ito, K.; Akashi, S.; Saito, B.; Katsuki, T. *Synlett* **2003**, 1809. (d) Taylor, M. S.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2004**, 126, 10558. (e) Taylor, A. M.; Schreiber, S. L. *Org. Lett.* **2006**, 8, 143. (f) Itoh, T.; Miyazaki, M.; Fukuoka, H.; Nagata, K.; Ohsawa, A. *Org. Lett.* **2006**, 8, 1295. (g) Miyazaki, M.; Ando, N.; Sugai, K.; Seito, Y.; Fukuoka, H.; Kanemitsu, T.; Nagata, K.; Odanaka, Y.; Nakamura, K. T.; Itoh, T. *J. Org. Chem.* **2011**, 76, 534. (h) Mittal, N.; Sun, D. X.; Seidel, D. *Org. Lett.* **2014**, 16, 1012. (i) Mons, E.; Wanner, M. J.; Ingemann, S.; van Maarseveen, J. H.; Hiemstra, H. *J. Org. Chem.* **2014**, 79, 7380.
- (25) Dhanasekaran, S.; Bisai, V.; Unhale, R. A.; Suneja, A.; Singh, V. K. *Org. Lett.* **2014**, 16, 6068.
- (26) (a) Fu, P.; Snapper, M. L.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2008**, 130, 5530. (b) Karmakar, R.; Suneja, A.; Bisai, V.; Singh, V. K. *Org. Lett.* **2015**, 17, 5650.
- (27) Isoindolinone **4i** can be obtained in 47% yield along with 25% of **4j** under refluxing CHCl₃ for 24 h.
- (28) (a) Phukan, P. *J. Org. Chem.* **2004**, 69, 4005. (b) Beisel, T.; Manolikakes, G. *Org. Lett.* **2013**, 15, 6046.
- (29) Patra, M.; Ingram, K.; Leonidova, A.; Pierroz, V.; Ferrari, S.; Robertson, M. N.; Todd, M. H.; Keiser, J.; Gasser, G. *J. Med. Chem.* **2013**, 56, 9192 and references cited.